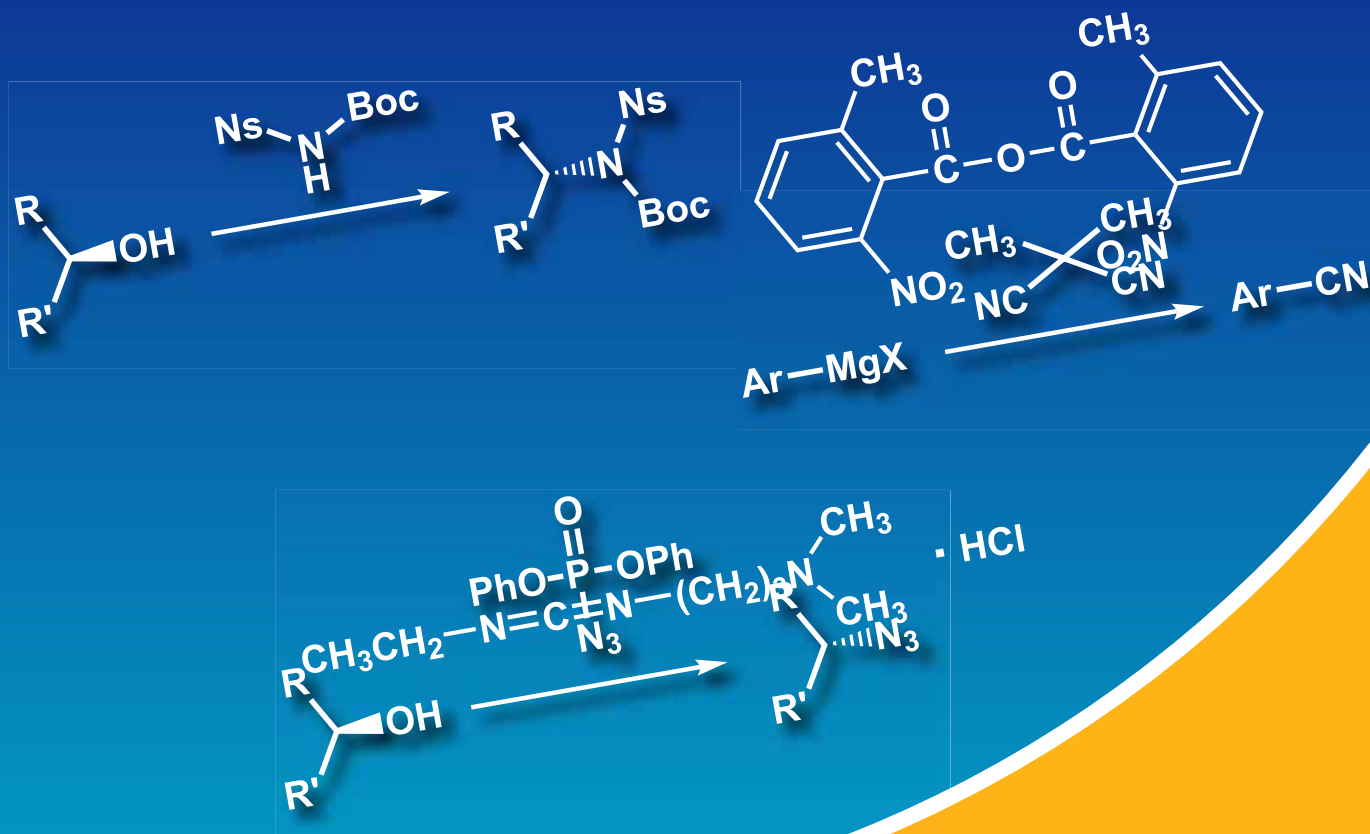


# C-N键形成反应

## C-N Bond Formation Reactions



胺化试剂

硝化试剂

氰化试剂

叠氮化和重氮化试剂

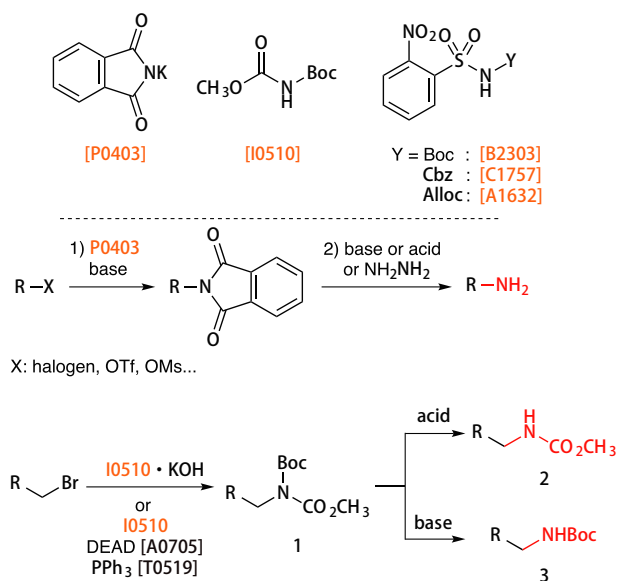
胍基化试剂

# C-N键形成反应

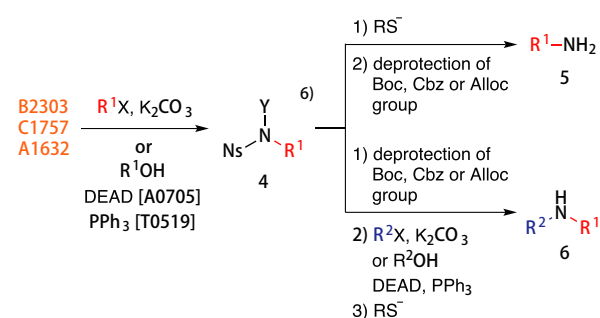
含氮化合物广泛存在于天然产物如氨基酸、核酸和生物碱到合成化合物如电子材料和聚酰胺中。此外，不同的含氮官能团如氨基和硝基具有不同的偶联反应式和氧化态，因此C-N键因官能团不同其成键反应也不同。由于C-N键形成反应包含了从简单引入官能团到构建杂环等多种方法，我们很难完全彻底介绍C-N键成键反应，因此本手册主要介绍TCl试剂通过C-N键形成反应构建含氮官能团。此外，手册中有些氧化剂可能不适用于C-N键形成反应，但是氰基可以转化为其他含氮官能团供参考。

## ● 胺化试剂

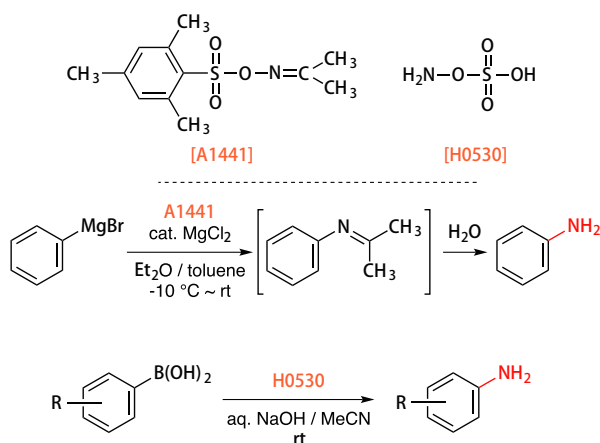
Gabriel胺合成法<sup>1)</sup>是目前广泛使用的亲核胺化反应，邻苯二甲酰亚胺钾以及相似的胺化试剂也已有报道。亚氨基二甲酸叔丁酯钾盐<sup>[I0510]</sup>与卤代烷反应生成酰亚胺<sup>2)</sup>。1) 分别经酸或碱处理得到N-甲氧基羰基胺<sup>2</sup>和N-Boc胺<sup>3</sup>。此外，<sup>I0510</sup>可用于Mitsunobu反应，将羟基转化为氨基<sup>3)</sup>。



与此同时，福山等人报道了用邻硝基苯磺酰基 (Ns) 保护的磺胺类化合物<sup>[B2303][C1757][A1632]</sup>合成胺的方法<sup>4)</sup>。这些物质分别在碱性条件下与卤代烷或在含醇条件下通过Mitsunobu反应得到含保护基的胺<sup>4</sup>。通过选择性地去除两个保护基团可得到伯胺<sup>5</sup>。然而，当保留<sup>4</sup>的Ns基团时，可以在烷基化后并相继去除保护基得到仲胺<sup>6</sup>。

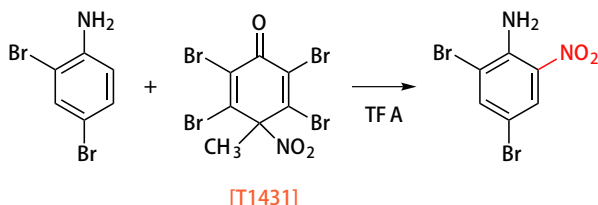


乙酰肟O-(2,4,6-三甲基苯磺酸盐)<sup>[A1441]</sup>可用作亲电胺化试剂<sup>6)</sup>。<sup>A1441</sup>在催化量的氯化镁下与格氏试剂反应可高产率生成伯胺<sup>7)</sup>。此外，羟胺-磺酸 (HSA) <sup>[H0530]</sup>也可用作胺化试剂。HSA作为氨基阳离子，通过苯硼衍生物处理可生成伯苯胺。该反应优点是不需要过渡金属催化剂就能进行反应，操作方便<sup>7)</sup>。



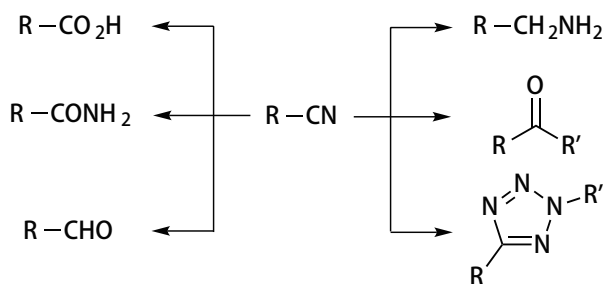
## ● 硝化试剂

硝基是强吸电子官能团，可以降低芳香环的电子密度，增加硝基烷 $\alpha$ -位质子的酸度。因此，含硝基的化合物可以发生独特的反应，如Henry反应和Nef反应。芳环的硝化一般是通过浓硝酸和浓硫酸进行亲电取代反应。而2,3,5,6-四溴-4-甲基-4-硝基-2,5-环己二酮[T1431]是一种温和的芳香族化合物硝化试剂<sup>9,10</sup>。例如，富含电子的芳香族化合物如苯胺，经T1431处理可发生邻位或对位硝化反应。

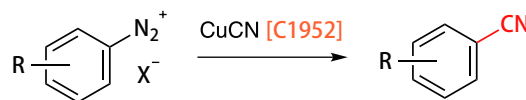


## ● 氰化试剂

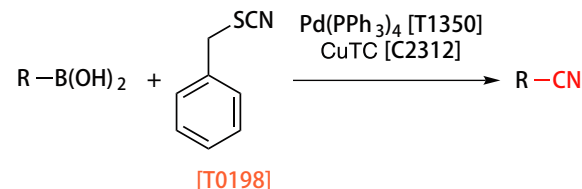
氰基是强吸电子基团，许多氰化试剂广泛用于有机合成中。氰基可以通过水解转化为其他官能团，例如羧酸或酰胺，也可通过一些还原剂还原为胺或醛。此外，还可以通过与格氏试剂或有机锂的亲核加成反应，将腈转化为相应的不对称酮。另一方面，氰基还可用于和其他多键化合物的环加成反应。例如，烷基腈与叠氮化物环合得到四唑。为了引入氰基，氰化钾[P1613]与烷基卤的反应是一种典型的合成方法，氰化铜介导的Sandmeyer反应和Rosenmund-von Braun反应也久负盛名。近年来发展了钯催化剂与一些氰化试剂的氰化反应。硫氰酸苄酯[T0198]<sup>11</sup>，氰基乙酸乙酯[C0441]<sup>12</sup>，丁基叔丁基[B1274]<sup>13</sup>和丙酮氰醇[M0361]<sup>14</sup>作为氰离子等价物。近来，一些研发小组报道了乙腈<sup>15</sup>和二甲基丙二腈<sup>16</sup>[D5514]可以用作氰基源。这样，有多种氰化试剂可用于不同物质的直接氰化反应。



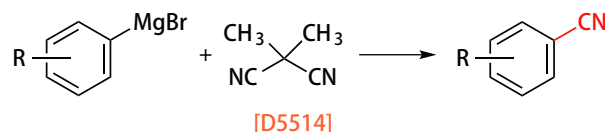
## Sandmeyer reaction



## Pd-Catalyzed cross-coupling cyanation

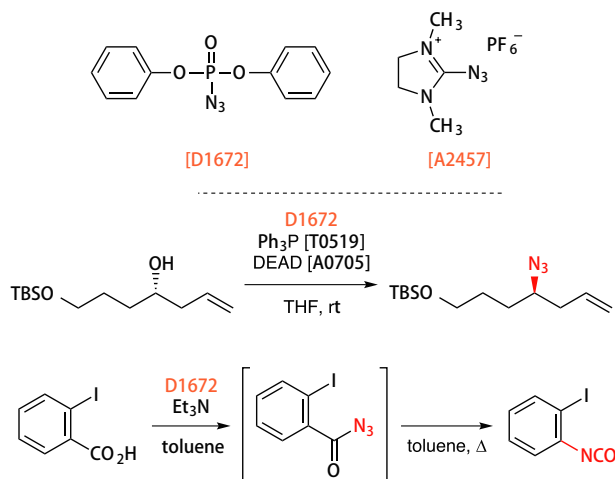


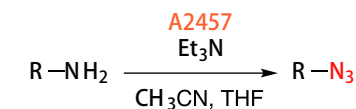
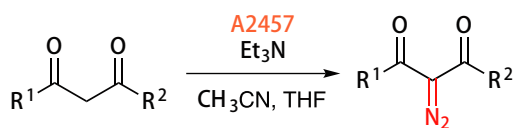
## Cyanation of Grignard reagent



## ● 叠氮化和重氮化试剂

有机叠氮化物可以通过叠氮化钠[S0489]与卤代烷基化合物反应，或与三氟甲磺酰基叠氮化物和伯胺的反应来合成。然而，这些叠氮化物源因具有高度爆炸性所以很难处理。相比之下，Shioiri等人开发了一种稳定易操作的叠氮化试剂DPPA [D1672]<sup>17</sup>。当在Mitsunobu条件下用DPPA处理乙醇时，可以高收率得到构型翻转的叠氮化合物。DPPA不仅用于Curtius重排，还可用作缩合试剂<sup>18</sup>。此外，Kitamura等人开发的2-叠氮基-1,3-二甲基咪唑啉六氟磷酸盐(ADMP) [A2457]是一种热稳定性高、对冲击和摩擦反应性低的结晶重氮转移试剂。ADMP可与多种伯胺反应生成叠氮化物<sup>19</sup>，同时可与1,3-二羰基化合物高产率生成 $\alpha$ -重氮化合物<sup>20</sup>。

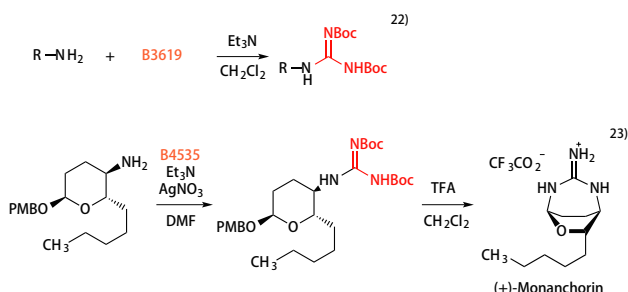
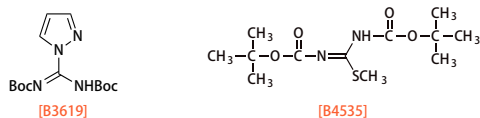




R: alkyl, aryl

## 胍基化试剂

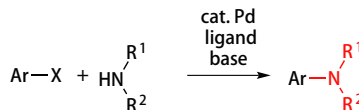
胍基是有机碱性中具有较强碱性的基团，存在于许多具有生物活性的化合物以及精氨酸中。例如，源于河豚毒素的石房蛤毒素和河豚毒素，以及海绵中作为HIV抑制剂的batzelladine A均为含有胍基的化合物。因此，胍基化试剂常被用于药物研发领域中胍类衍生物的合成<sup>21)</sup>。胍基的引入主要是通过氨基的加成反应获得。



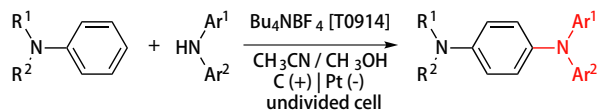
## 通过交叉偶联反应形成C-N键

Buchwald<sup>24)</sup>和Hartwig<sup>25)</sup>分别独立报道了在钯催化剂和强碱存在下，胺和芳基卤化物的一种新型偶联反应。这个反应称为Buchwald-Hartwig交叉偶联反应。该反应将氨基引入芳族化合物，也可用于构建含氮杂芳环<sup>26)</sup>。因此，该反应广泛应用于天然产物的全合成、药物化学和过程化学。最近报道了一种新的电氧化C-N交叉偶联反应<sup>27)</sup>。该反应具有以下优点：对位选择性交叉偶联反应；无金属催化剂反应；该反应唯一的副产物是通过C-H键活化产生的氢气，因此被称为绿色反应。

### Buchwald-Hartwig cross-coupling



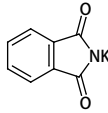
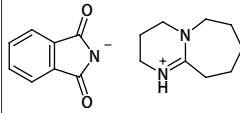
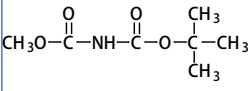
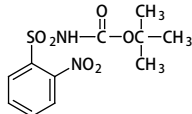
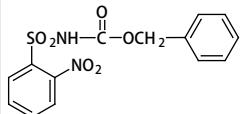
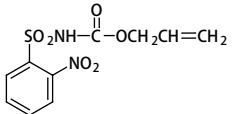
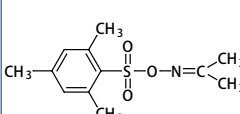
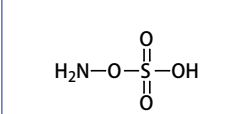
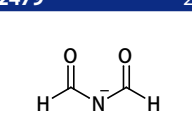
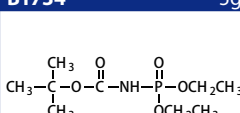
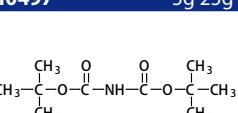
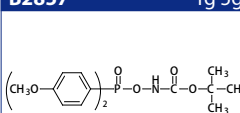
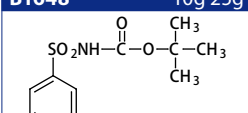
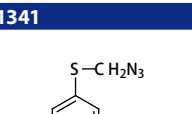
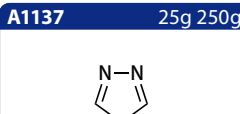
### Electrooxidative C-N cross-coupling



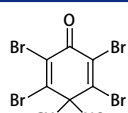
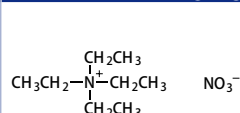
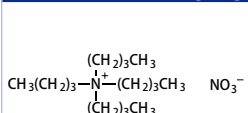
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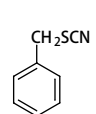
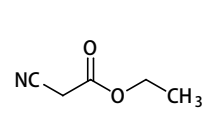
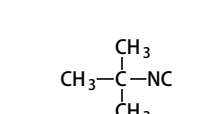
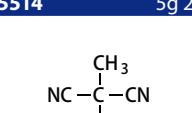
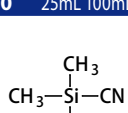
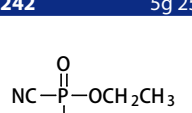
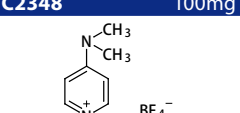
## 胺化试剂 Reagents for Amination

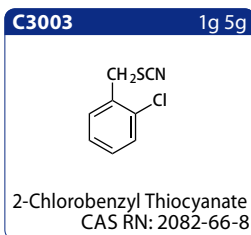
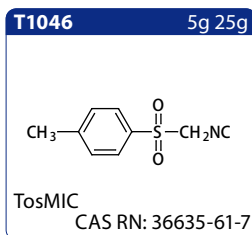
		<b>P0403</b> 25g 500g  Phthalimide Potassium Salt CAS RN: 1074-82-4	<b>P1235</b> 25g  Phthalimide DBU Salt CAS RN: 119812-51-0	<b>I0510</b> 5g  <i>tert</i> -Butyl Methyl Iminodicarboxylate CAS RN: 66389-76-2
<b>B2303</b> 1g 5g 25g  <i>N</i> -Boc-2-nitrobenzenesulfonamide CAS RN: 198572-71-3	<b>C1757</b> 5g 25g  <i>N</i> -Cbz-2-nitrobenzenesulfonamide CAS RN: 245365-64-4	<b>A1632</b> 5g  <i>N</i> -Alloc-2-nitrobenzenesulfonamide CAS RN: 90916-29-3	<b>A1441</b> 5g 25g  Acetoxime <i>O</i> -(2,4,6-Trimethylphenylsulfonate) CAS RN: 81549-07-7	<b>H0530</b> 25g 100g 500g  Hydroxylamine- <i>O</i> -sulfonic Acid CAS RN: 2950-43-8
<b>D2479</b> 25g  Sodium Diformylamide CAS RN: 18197-26-7	<b>B1734</b> 5g  <i>N</i> -Boc-phosphoramidic Acid Diethyl Ester CAS RN: 85232-02-6	<b>I0497</b> 5g 25g  Di- <i>tert</i> -butyl Iminodicarboxylate CAS RN: 51779-32-9	<b>B2857</b> 1g 5g  <i>tert</i> -Butyl [Bis(4-methoxyphenyl)-phosphinyloxy]carbamate CAS RN: 619333-95-8	<b>B1648</b> 10g 25g  <i>N</i> -Boc- <i>p</i> -toluenesulfonamide CAS RN: 18303-04-3
<b>A1341</b> 5g  Azidomethyl Phenyl Sulfide CAS RN: 77422-70-9	<b>A1137</b> 25g 250g  4-Amino-1,2,4-triazole CAS RN: 584-13-4			

## 硝化试剂 Reagents for Nitration

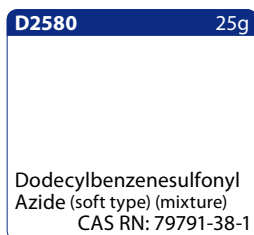
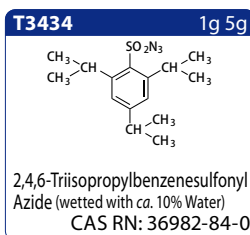
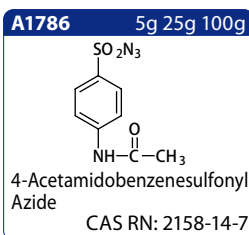
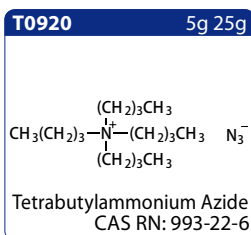
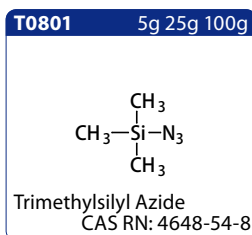
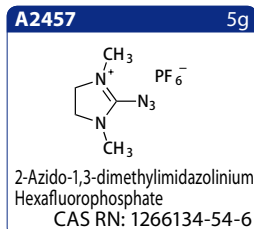
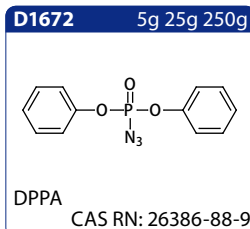
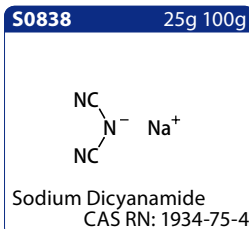
<b>T1431</b> 5g 25g  2,3,5,6-Tetrabromo-4-methyl-4-nitro-2,5-cyclohexadien-1-one CAS RN: 95111-49-2	<b>T3082</b> 5g 25g  Tetraethylammonium Nitrate CAS RN: 1941-26-0	<b>T3651</b> 5g 25g  Tetrabutylammonium Nitrate CAS RN: 1941-27-1
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## 氰化试剂 Reagents for Cyanation

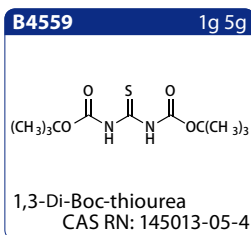
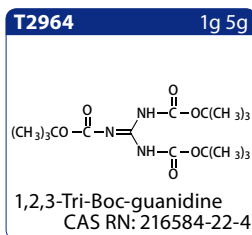
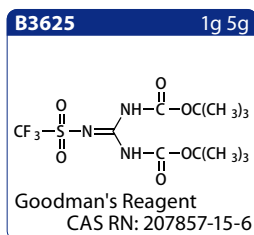
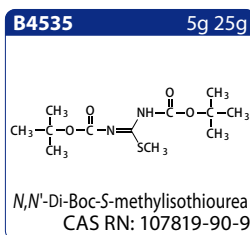
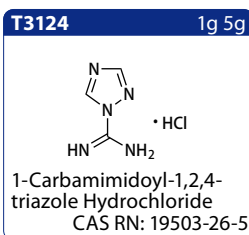
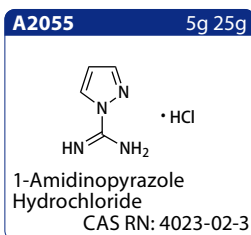
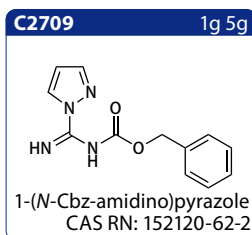
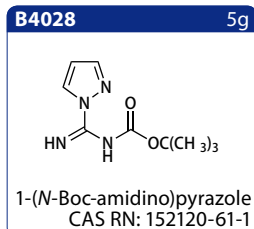
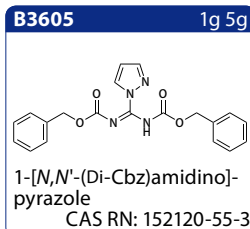
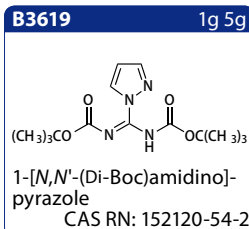
<b>T0198</b> 25g 500g  Benzyl Thiocyanate CAS RN: 3012-37-1	<b>C0441</b> 25g 500g  Ethyl Cyanoacetate CAS RN: 105-56-6	<b>B1274</b> 5mL 25mL  <i>tert</i> -Butyl Isocyanide CAS RN: 7188-38-7
<b>D5514</b> 5g 25g  Dimethylmalonitrile CAS RN: 7321-55-3	<b>C1952</b> 25g 300g CuCN Copper(I) Cyanide CAS RN: 544-92-3	<b>T0990</b> 25mL 100mL 500mL  Trimethylsilyl Cyanide CAS RN: 7677-24-9
<b>C1242</b> 5g 25g  Diethyl Cyanophosphonate CAS RN: 2942-58-7	<b>C2348</b> 100mg  1-Cyano-4-(dimethylamino)pyridinium Tetrafluoroborate CAS RN: 59016-56-7	



## 叠氮化和重氮化试剂 Reagents for Azidation and Diazotization



## 胍基化试剂 Reagents for Guanidinylation







## **梯希爱(上海)化成工业发展有限公司**

试剂热线：800-988-0390或021-67121386

大包装热线：800-988-1865

传真：021-67121385

邮箱：Sales-CN@TCIchemicals.com

地址：上海化学工业区普工路96号

邮编：201507

**[www.TCIchemicals.com](http://www.TCIchemicals.com)**